

Ecosystem Consequences of Changing Inputs of Terrestrial Dissolved Organic Matter to Lakes: Current Knowledge and Future Challenges

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ABSTRACT

Lake ecosystems and the services that they provide to people are profoundly influenced by dissolved organic matter derived from terrestrial plant tissues. These terrestrial dissolved organic matter (tDOM) inputs to lakes have changed substantially in recent decades, and will likely continue to change. In this paper, we first briefly review the substantial literature describing tDOM effects on lakes and ongoing changes in tDOM inputs. We then identify and provide examples of four major challenges which limit predictions about the implications of tDOM change for lakes, as follows: First, it is currently difficult to forecast future tDOM inputs for particular lakes or lake regions. Second, tDOM influences ecosystems via complex, interacting, physical-chemical-biological effects and our

holistic understanding of those effects is still rudimentary. Third, non-linearities and thresholds in relationships between tDOM inputs and ecosystem processes have not been well described. Fourth, much understanding of tDOM effects is built on comparative studies across space that may not capture likely responses through time. We conclude by identifying research approaches that may be important for overcoming those challenges in order to provide policy- and management-relevant predictions about the implications of changing tDOM inputs for lakes.

Key words: lake; ecosystem; dissolved organic matter; dissolved organic carbon; terrestrial inputs; allochthonous; environmental change; review.

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INTRODUCTION

Terrestrially derived dissolved organic matter (terrestrial DOM or tDOM) is increasingly recognized as a fundamental control on lake ecosystem structure

and function (Jones 1992; Kullberg and others 1993; Williamson and others 1999; Prairie 2008). Like other major drivers such as nutrient concentrations and fishing pressure, tDOM has far-reaching effects on freshwater ecosystems. These effects occur at multiple levels of biological organization, ranging from cellular chemical stress to ecosystem biogeochemical cycles (Steinberg and others 2006; Prairie 2008). Furthermore, tDOM concentrations vary widely across the landscape, and inputs of tDOM to surface waters have changed substantially over the past several decades in many north temperate and boreal regions (Hanson and others 2007; Monteith and others 2007; Figure 1).

Given these two observations—that tDOM concentrations fundamentally shape lake ecosystems, and that these concentrations are changing through time—what are the implications for the future structure and function of these systems? Despite our considerable understanding of the role of tDOM in lakes, this is a surprisingly difficult question to answer. In this paper, we briefly review the effects of tDOM on lake ecosystems and the causes of recent changes in tDOM concentrations. We draw from several excellent reviews as well as more recent work, integrating perspectives from watershed hydrology, physical limnology, biogeochemistry, microbial and food web ecology, and other fields. We then focus on exploring four challenges that currently make predictions about the implications of changing tDOM inputs difficult, and conclude by suggesting some research avenues that may help to overcome these challenges.

WHAT IS tDOM AND HOW DOES IT ENTER LAKES?

Terrestrial DOM includes a diverse and variable suite of substances that originate from the tissues of terrestrial plants, is typically modified in the soil environment, and is ultimately transported to lakes by groundwater and surface water. Plant materials, including structural compounds like cellulose and lignin, are modified by interactions with minerals and microorganisms in the soil environment, where conditions like pH, temperature, and redox potential regulate solubility and rates of decomposition (Thurman 1985). The interaction of these physical and biological processes alters the chemical composition of soil organic matter, yielding a mixture of substances of diverse molecular size, age, and biological availability (Neff and Asner 2001). Although some of these substances are mineralized or sequestered in the soil, a substantial

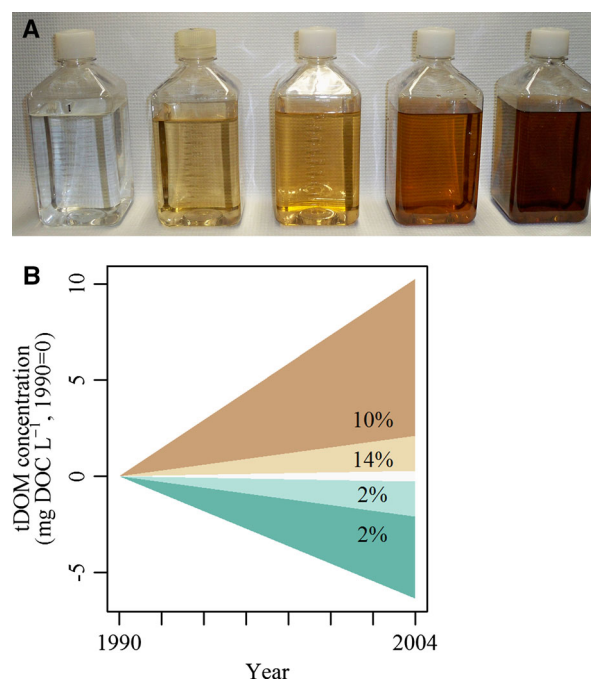


Figure 1. **A** The differences in color between these samples of water from five different lakes are due to different concentrations of terrestrially derived dissolved organic matter (tDOM). **B** Concentrations of tDOM in surface waters have changed over the past several decades in many surface waters. Numbers show the proportion of 500 lakes in which the tDOM trends from 1990 to 2004 falls within the shaded region. Increases greater than $0.02 \text{ mg DOC L}^{-1} \text{ y}^{-1}$ were observed in 24% of lakes. Redrawn from Monteith and others (2007) using data from the ICP Waters program (adapted by permission from Macmillan Publishers Ltd: Monteith and others 2007, copyright 2007).

portion can be exported to aquatic systems. This export from the watershed ranges from 1 to $10 \text{ g C m}^{-2} \text{ y}^{-1}$ or higher, depending on the ecosystem (Mulholland 2003), and globally constitutes about half of terrestrial net ecosystem production on an annual basis (Battin and others 2009). Hydrology is an important control on export, both as the vehicle for transporting organic matter and because wet soils accumulate organic matter faster than they mineralize it and so have more available for export (Freeman and others 2001a). Export may vary among nearby watersheds due to differences in hydrology or in other factors such as terrestrial NPP, watershed size, or the areal extent of wetlands and their proximity to surface waters (Gergel and others 1999; Canham and others 2004; Jansson and others 2008).

The diverse chemical composition of tDOM, and the spatial and temporal variability in that composition, makes it difficult to fully and simply

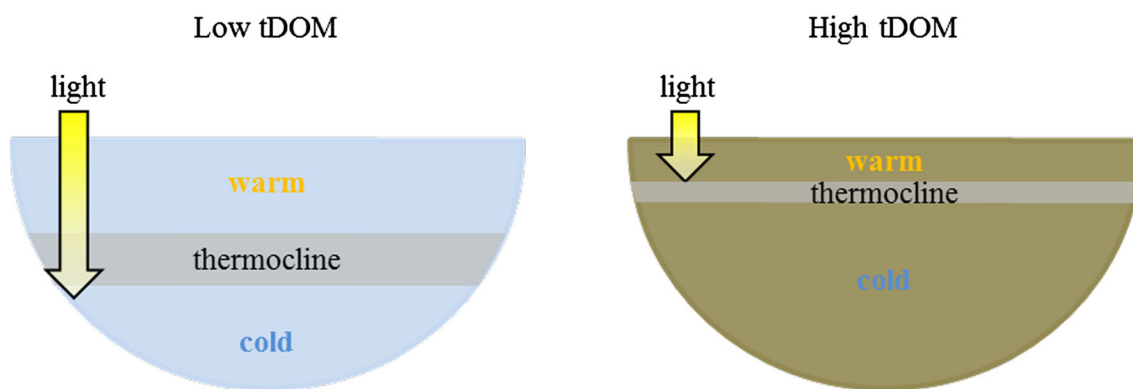


Figure 2. Light is extinguished rapidly with depth in a high-tDOM lake with dark water (*right*), and much less rapidly in a low-tDOM lake (*left*). Consequently, the water volume and bottom area capable of supporting photosynthesis are much lower in the high-tDOM lake. High-tDOM lakes also have steeper and shallower thermal gradients between warm oxygenated surface water and cold, potentially deoxygenated deep water. These differences strongly impact metabolic rates, biogeochemical processes, and animal habitat.

characterize (Sleighter and Hatcher 2007; Minor and others 2014). In general, however, a large fraction of the terrestrial DOM that is exported to aquatic ecosystems is comprised of humic substances (that is, humic and fulvic acids) that contain aromatic hydrocarbons including phenols, carboxylic acids, quinones, and catechol (McDonald and others 2004). These molecules are relatively resistant to microbial degradation by virtue of their molecular structure and high C:N and C:P ratios (McKnight and Aiken 1998). They also absorb light strongly in the ultraviolet and short wavelength visible region of the spectrum, giving water a brown, tea-stained color that affects light and heat penetration (Jones 1992). Operationally, DOM concentrations are often measured in terms of dissolved organic carbon, or DOC.

A FUNDAMENTAL CONTROL ON LAKE ECOSYSTEM STRUCTURE AND FUNCTION

Two properties of the complex suite of molecules that comprise tDOM have major implications for the structure of lake ecosystems.

First, tDOM absorbs solar radiation at particular wavelengths, changing the vertical distribution of light and heat (Kirk 1994; Fee and others 1996) (Figure 2). Light and temperature control metabolic rates, primary productivity, biogeochemistry, the distribution of organisms, and a host of other processes in lakes. In many lakes, terrestrial DOM is the primary regulator of water column transparency to the portion of shortwave energy (visible and ultraviolet light) that penetrates the

near-surface layer (Morris and others 1995; Williamson and others 1996). All else being equal, higher tDOM concentrations drive faster light extinction and a vertical distribution of heat that is more heavily weighted toward the surface, unless there is sufficient mixing energy (most often in the form of wind shear) to prevent stratification (Pérez-Fuentetaja and others 1999; Houser 2006). Faster light extinction limits light availability to primary producers and alters interactions between visual predators and their prey. Warmer surface water results in stronger outward energy fluxes, so high-tDOM lakes are generally colder overall (Tanentzap and others 2008; Read and Rose 2013). The surface-weighted distribution of heat in low-transparency lakes also means that thermal stratification occurs closer to the surface and tends to be more stable (Kling 1988; Read and Rose 2013; Palmer and others 2014); more stable stratification reduces the amount of vertical mixing and alters vertical gradients of dissolved oxygen and other chemicals (Imberger 1998; Wüest and Lorke 2003; MacIntyre and others 2006). This affects biogeochemical reaction rates and habitat suitability for aerobic organisms.

Second, loads of tDOM are an energetic input to the base of the lake food web and can support catabolic and anabolic metabolism (del Giorgio and Peters 1994; Pace and others 2004) (Figure 3). A portion of the load occurs as low-molecular weight compounds that can be rapidly consumed by heterotrophic bacteria (Berggren and others 2010). More recalcitrant, high molecular weight compounds comprise the majority of the load, but even these are slowly degraded and consumed if

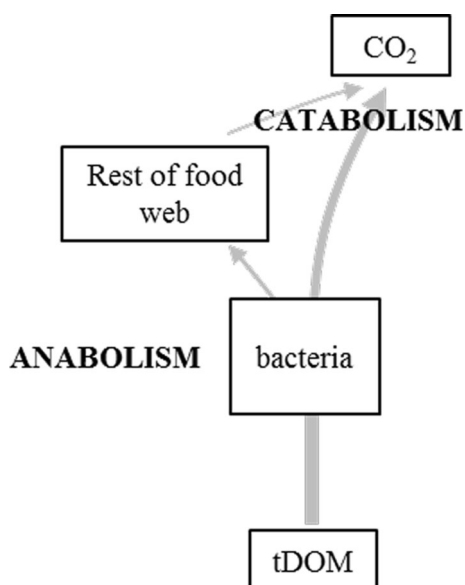


Figure 3. Terrestrial DOM provides a substrate for anabolic and catabolic metabolism, helping to support lake food webs and influencing carbon emissions to the atmosphere.

residence times are sufficient (Moran and Hodson 1990; Tranvik 1990; Volk and others 1997; Tranvik 1998; Aitkenhead-Peterson and others 2003; Young and others 2005). Photochemical reactions can also modify the lability of tDOM inputs (Geller 1986; Tranvik and Bertilsson 2001). Terrestrial DOM consumed by heterotrophic bacteria follows one of two pathways. Some is incorporated into cellular structures and thus becomes available to higher consumers like zooplankton and fishes, which may derive substantial portions of their biomass from terrestrial sources in lakes with large tDOM inputs (Grey and others 2001; Karlsson and others 2003; Pace and others 2004; Matthews and Mazumder 2006; Taipale and others 2008; Brett and others 2009; Cole and others 2011; Solomon and others 2011; Tanentzap and others 2014). Some is respired as CO_2 , adding to the pool of inorganic carbon dissolved in lake water. This contributes to a net flux of CO_2 from the water to the atmosphere in many lakes with large tDOM inputs (Hope and others 1996; Sobek and others 2003; Larsen and others 2011b). These contributions of tDOM to metabolic processes in lakes are variable but sometimes quite large.

TERRESTRIAL LOADS TO AQUATIC SYSTEMS ARE CHANGING

Terrestrial DOM loads and concentrations have increased over the past several decades in many

north temperate and boreal surface waters, in a phenomenon sometimes referred to as “browning” (Skjelkvale and others 2005; Roulet and Moore 2006; Kritzberg and Ekstrom 2012; SanClements and others 2012; Figure 1B). Temporal trends vary among systems, and include patterns of stable or decreasing systems, and include patterns of stable or decreasing DOC concentrations. For instance, Schindler and others (1997) observed decreasing DOC in a set of lakes experiencing long-term drought. Nonetheless, across broad regional and intercontinental scales, the majority of systems have experienced increases in DOC concentrations (Monteith and others 2007; Winterdahl and others 2014). For instance, trends in DOC from 1990 to 2004 were positive in 70% of 522 surveyed waters in North America and Europe, and DOC concentrations over roughly this period increased by 91% on average in monitored streams and lakes in the United Kingdom (Evans and others 2005; Monteith and others 2007).

A number of mechanisms related to climate change, atmospheric deposition, hydrology, and other drivers have been proposed as contributors to observed changes in DOC, and their relative importance has been debated (Evans and others 2006; Roulet and Moore 2006; Clark and others 2010). These mechanisms include (1) factors influencing the quantity and quality of plant-derived soil organic matter, such as climate effects on terrestrial net primary productivity and vegetation communities and nitrogen deposition effects on the characteristics of plant-derived organic matter and belowground C and N processing (Pregitzer and others 2004; Larsen and others 2011a); (2) factors influencing the solubility of soil organic matter, such as the impact of sulfate deposition on soil chemistry and the impact of temperature on extracellular enzyme activity in peat soils (Freeman and others 2001a; Clark and others 2005; De Wit and others 2007; Monteith and others 2007; Erlandsson and others 2008); and (3) factors influencing the hydrologic transport of tDOM to surface waters, including inter-annual or decadal-scale variation in precipitation and runoff patterns (Hongve and others 2004; Erlandsson and others 2008; Haaland and others 2010). Current synthesis suggests that many of these mechanisms play a role at certain spatial and temporal scales, but that the primary driver for decadal-scale increases, where observed, is linked to decreases in atmospheric sulfate deposition as a result of emissions regulations in North America and Europe (Monteith and others 2007; Erlandsson and others 2008; Clark and others 2010). These decreases seem to be changing soil chemistry in ways that increase the solubility of

tDOM (Clark and others 2005; De Wit and others 2007).

HOW WILL CHANGING tDOM LOADS AFFECT LAKES?

It is clear that important ecosystem processes as diverse as carbon cycling, fish production, and drinking water provisioning could be strongly impacted by changes in tDOM inputs. Yet as we describe in this section, despite the considerable existing literature, four fundamental gaps in our understanding make concrete predictions about such ecosystem responses challenging.

Challenge 1: Uncertainty About Future tDOM Loads

The impact on lake ecosystems of any future changes in tDOM inputs will depend first and foremost on the magnitude of those changes. The tDOM increases observed over the past few decades have been substantial in some regions, as described above. Future changes as a result of shifts in sulfate deposition and climate could also be substantial, but our ability to forecast those changes is currently limited.

Changes in sulfate deposition will affect tDOM loads at very broad spatial scales. In North America and Europe, deposition reductions will likely cease to be a major driver of increasing loads as legislated emission targets are met and soils recover. In fact, in these regions, the observed changes in tDOM concentrations over recent decades may represent recovery to a more pre-industrial state, rather than a novel disturbance; continued development of paleolimnological techniques for inferring past DOC concentrations could help address this question (Rouillard and others 2011; Brag e and others 2013). Conversely, in industrializing countries rapidly increasing emissions may drive decreases in tDOM loads in downwind regions with acid-sensitive soils. Soil heterogeneity, patchiness of emissions sources, atmospheric transport mechanisms, and other factors will create local heterogeneity in these broad-scale patterns.

Superimposed on these effects, climate change will increasingly alter the processes that drive tDOM loading. The net effect of climate change on tDOM loads in a particular location or region is difficult to predict, given the complexity of the processes that generate and transport tDOM and the potential for effects at time scales ranging from years to centuries (Table 1). For instance, warmer temperatures will favor not only greater inputs to

the soil OM pool via increased terrestrial primary production but also greater removals from that pool via increased soil respiration (Wu and others 2011), and the transport of that OM to lakes in the form of tDOM will vary in both quantity and quality depending on precipitation patterns and hydrology. Hydrologic change may also alter transport of iron, which, like tDOM, contributes to water color; given that many of the effects of high-tDOM water are due to its color, this potential change in the color of water with a given tDOM concentration may be important (Weyhenmeyer and others 2014). Mechanistic and phenomenological watershed models can forecast the net effects of these changes, and there is a clear need to continue developing, testing, and integrating these models with climate and vegetation projections (Futter and others 2007; Larsen and others 2011a).

Other regional- or global-scale environmental changes may also have an impact on tDOM loads and surface water browning. Notably, nitrate deposition may continue to affect tDOM loads via both its plant fertilization and soil acidification effects. Land-use and land-cover changes have also been implicated in altering tDOM fluxes (Mattsson and others 2005), and will interact with changing atmospheric deposition and climate conditions to regulate tDOM loading to lakes in a given region (for example, Winterdahl and others 2014).

Overall, specific quantitative or even qualitative predictions about future tDOM loads at relevant regional or lake-level spatial scales are difficult given our current understanding. Nonetheless, it seems likely that tDOM loads will continue to change in coming decades as anthropogenic effects reshape soil organic matter pools and their connections to aquatic systems. Given the potential for these changes to profoundly influence lake ecosystems, there is a concurrent need for aquatic ecologists to consider the potential impacts of changing tDOM loads.

Challenge 2: Complex Interacting Effects

Terrestrial DOM influences lake biota and biogeochemistry directly, and also indirectly via its regulating effects on the physical environment. These features create complex networks of interactions among physics, biology, and chemistry. For simplicity, we consider the effects of tDOM change but not potential interactions with climate or other ongoing environmental changes (Kritzberg and others 2014; Weidman and others 2014). We focus our discussion in this section and those that follow on predicting the effect of increases in tDOM

Table 1. Some Mechanisms By Which Changes in Climate (Temperature, Precipitation, and Hydrology) Have Affected, and May Continue to Affect, Terrestrially Derived Dissolved Organic Matter (tDOM) Loads to Lakes at Time Scales Ranging From Years to Centuries

Driver	Mechanism	Effect on tDOM load	Reference
<i>Years</i>			
Increased temperature	Increased soil decomposition rate, decreased soil OM pool	Decrease	Kirschbaum (2006)
Increased temperature	Increased microbial release of sorbed soil OM	Increase	Freeman and others (2001a), (2001b), von Lutzow and Kogel-Knabner (2009)
Increased temperature	Increased soil oligochaete activity	Increase	Cole and others (2002)
Increased precipitation	Increased GPP of terrestrial vegetation, increased soil OM pool	Increase	Wu and others (2011)
Increased runoff	Increased tDOM transport through catchment	Increase	Tranvik and Jansson (2002), Pastor and others (2003), Erlandsson and others (2008)
Drought	Decreased tDOM transport through catchment	Decrease	Schindler and others (1997)
More flashy runoff	More flashy transport	More flashy	Schindler and others (1997), Hongve and others (2004)
Increased frequency or magnitude of drought-rewetting cycles	Increased aerobic mineralization of peat, coupled with flushing out of soluble tDOM	Increase?	McDonald and others (1991), Mitchell and McDonald (1992), Hughes and others (1998), Clark and others (2005), (2009)
Change in snow cover duration	Soil frost depth, soil solution tDOM concentrations and fluxes	Change	Haei and others (2010)
<i>Decades</i>			
Increased temperature	Terrestrial vegetation assemblage shifts towards species with greater GPP and biomass, increased soil OM pool	Increase	Barichivich and others (2013)
Increased temperature	Melting permafrost, increase or decrease in tDOM load depending on depth of organic soil layer relative to hydrologic flowpaths	Change	Striegl and others (2005), Frey and McClelland (2009)
Increased temperature and altered hydrology	Change in wetland and peatland soil OM stocks	Change	Davidson and Janssens (2006)
<i>Centuries</i>			
Increased temperature and altered geographical distribution of precipitation	Change in wetland and peatland equilibrium states, change in geographical distribution of wetlands	Change	Belyea and Malmer (2004)

GPP, gross primary production; OM, organic matter.

(“browning”), but in general our predictions can be reversed for scenarios of decreasing tDOM.

Lake carbon cycles provide one example of these complex effects. Lakes are hotspots for carbon processing on the landscape, and play a significant role in regional and global carbon cycles (Cole and others 2007; Tranvik and others 2009). In general, tDOM concentration is positively correlated with ecosystem respiration and CO₂ release to the atmosphere (del Giorgio and Peters 1994; Sobek and others 2005; Solomon and others 2013). Yet while that relationship is a powerful heuristic, the com-

plexity of the underlying mechanisms adds considerable noise and limits its utility for predicting the carbon balance implications of changing tDOM inputs. For instance, consider the ways in which tDOM interacts with phytoplankton production, which removes CO₂ from the water via photosynthetic fixation (Figure 4A). Terrestrial DOM reduces light availability at a given depth, which can limit production, although it also absorbs heat and decreases the depth of the mixed layer, constraining epilimnetic phytoplankton to the near-surface zone where light availability is higher than it is at

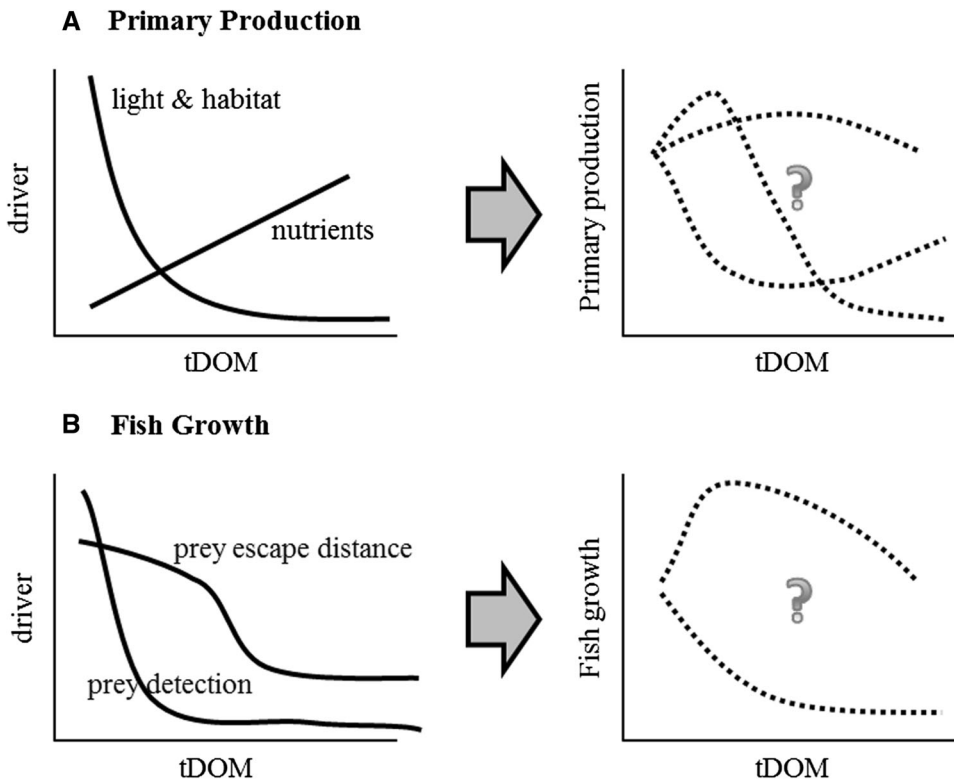


Figure 4. Terrestrial DOM affects lake processes through complex networks of physical, chemical, and biological effects. Predicting the implications of tDOM change, therefore, requires research that is both mechanistic and holistic. Interacting and counteracting effects of tDOM on light, nutrients, and other factors make it difficult to predict how tDOM change will affect ecosystem services like primary production (A) and fish growth (B).

greater depths (Jones 1992; Carpenter and others 1998). Mineralization of tDOM by heterotrophic bacteria releases nutrients and inorganic carbon, both of which may stimulate phytoplankton production; on the other hand, the bacteria themselves may outcompete phytoplankton for the nutrients that are released, and stronger, shallower stratification limits phytoplankton access to internally recycled nutrients (Jackson and Hecky 1980; Jones 1992; Hessen 1998; Jansson and others 2012). In short, tDOM has a series of cascading and interacting physical, chemical, and biological effects that strongly influence rates of primary production; other carbon cycle processes are similarly influenced, and also interact with each other and with rates of primary production (Brothers and others 2014). These complexities are not unique to the carbon cycle; for instance, tDOM-induced changes in light regime can alter benthic nitrogen sinks by changing the redox conditions that control nitrification and denitrification (Fork and Heffernan 2013).

Behavior adds another layer of complexity in considering tDOM effects on animals such as fishes (Figure 4B). Fish are keystone species in many lakes, and support culturally and economically valuable fisheries. Processes that drive the fitness of an individual fish and the dynamics of fished

populations—such as avoiding predators, capturing food, growing, and reproducing—are strongly influenced by tDOM concentrations (Williamson and others 1999; Stasko and others 2012). Dark water reduces the abundance of the zooplankton and zoobenthos that form the base of food chains supporting fishes (Karlsson and others 2009; Jones and others 2012; Kelly and others 2014). It also can drive predator-prey interactions by favoring species adapted to feed in low-light environments. For example, perch feeding on zooplankton are at a competitive disadvantage relative to roach in high-tDOM, low-light conditions (Estlander and others 2010). This effect is exacerbated where the tDOM-driven light limitation reduces the abundance of macrophytes (Sondergaard and others 2013), which provide refuge habitat that normally lowers predation risk and increases invertebrate prey availability for perch (Olin and others 2010). Although warmer surface waters in dark lakes could enhance growth rates of some fish species, the accompanying steeper thermal stratification promotes hypolimnetic hypoxia, decreasing available fish habitat. Spawning habitat availability and suitability similarly depend on temperature and dissolved oxygen. Fish behavior responds to all of these forces, as individuals try to maximize fitness by allocating activities like foraging in time and

space, triggering cascading effects on the abundance or behavior of lower trophic levels which in turn feed back to fish.

Challenge 3: Non-Linear Relationships and Context Dependence

Many processes in lakes are non-linearly related to tDOM concentration. This has two important implications. First, a given change in tDOM load will have different effects depending on the initial context. Second, identifying the shapes of relationships between lake processes and tDOM, and the position of threshold or inflection points in those relationships, is an important research goal.

Changes in light availability and thermal structure in response to differences in tDOM concentrations (for example, Figure 2) are one important example of a non-linear effect. A given change in tDOM concentration alters the heat distribution and stratification strength considerably in a lake that was initially low-tDOM but only slightly in a lake that was initially high-tDOM (Snucins and Gunn 2000). Similarly, Read and Rose (2013) modeled lakes across a DOC concentration gradient and found striking non-linearity in thermal responses to climate, with clearer lakes being increasingly sensitive to climate variability and DOC perturbations. The power-law relationship between light attenuation and tDOM concentration (Morris and others 1995) structures these non-linear physical patterns. These effects are likely stronger in small lakes than in large ones, because wind-driven mixing and basin-scale hydrodynamics become more important in larger lakes, reducing the relative importance of water clarity as a control on thermal structure (Fee and others 1996; Read and others 2012). The non-linear effects of light and heat propagate up the food web to control biomass production at trophic levels ranging from primary producers to top consumers, including benthic algae, zoobenthos, zooplankton, and fishes (Ask and others 2009; Karlsson and others 2009; Finstad and others 2014; Kelly and others 2014). Many of these food web studies suggest a threshold DOC concentration of roughly 10 to 14 mg l⁻¹ above which consumer production is severely reduced, and a recent experimental pond study that found little effect on zoobenthos as a result of raising DOC concentrations to 10 mg l⁻¹ lends further credence to this idea (Jonsson and others 2015). Yet the existence of such a threshold and the DOC concentration at which it occurs has not been explicitly investigated.

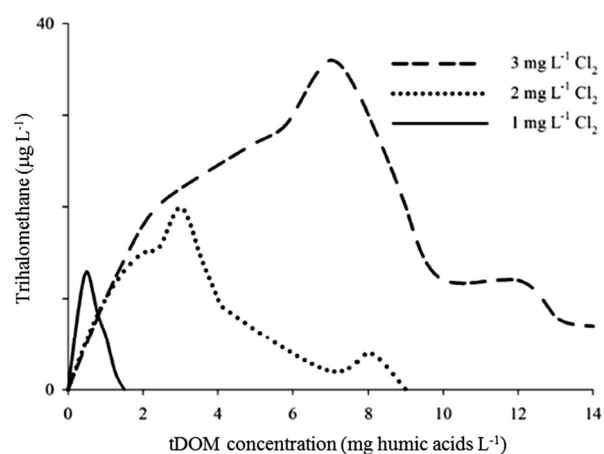


Figure 5. Terrestrial DOM concentrations are non-linearly related to many important ecosystem processes. Describing the shape of these relationships, including critical thresholds of tDOM, is therefore an important objective. The production of carcinogenic trihalomethanes during drinking water purification varies non-linearly with tDOM concentration, and with other factors such as dosage of chlorine (Cl₂). Redrawn from Adin and others (1991), copyright 1991, with permission from Elsevier.

The effects of changing tDOM loads on drinking water supplies will also be context-dependent (Figure 5). Lakes and other surface waters provide drinking water for a substantial portion of the world population, including, for instance, 56% of the United States population (Kenny and others 2009). Elevated tDOM concentrations can increase treatment costs and reduce quality of drinking water in several ways, as they can cause color, taste, and odor problems; transport heavy metals and organic pollutants; promote the growth of bacteria; and produce chlorination byproducts called trihalomethanes, which may be carcinogenic (Matilainen and others 2010; Ledesma and others 2012). In general, trihalomethane production increases with DOC concentration, but the mechanism is complex and depends on chlorine dosage, bromine concentration, pH, and temperature (Adin and others 1991; Krasner and others 1994). These concerns are motivating modified treatment strategies such as coagulation, magnetic ion exchange, and adsorbents, typically increasing costs for utilities and consumers (Matilainen and Siljanpaa 2010).

Challenge 4: Substituting Space for Time

Surveys of lakes that differ in tDOM concentration have been instrumental in developing understanding of tDOM effects (Karlsson and others 2009; Lapierre and others 2013). For example,

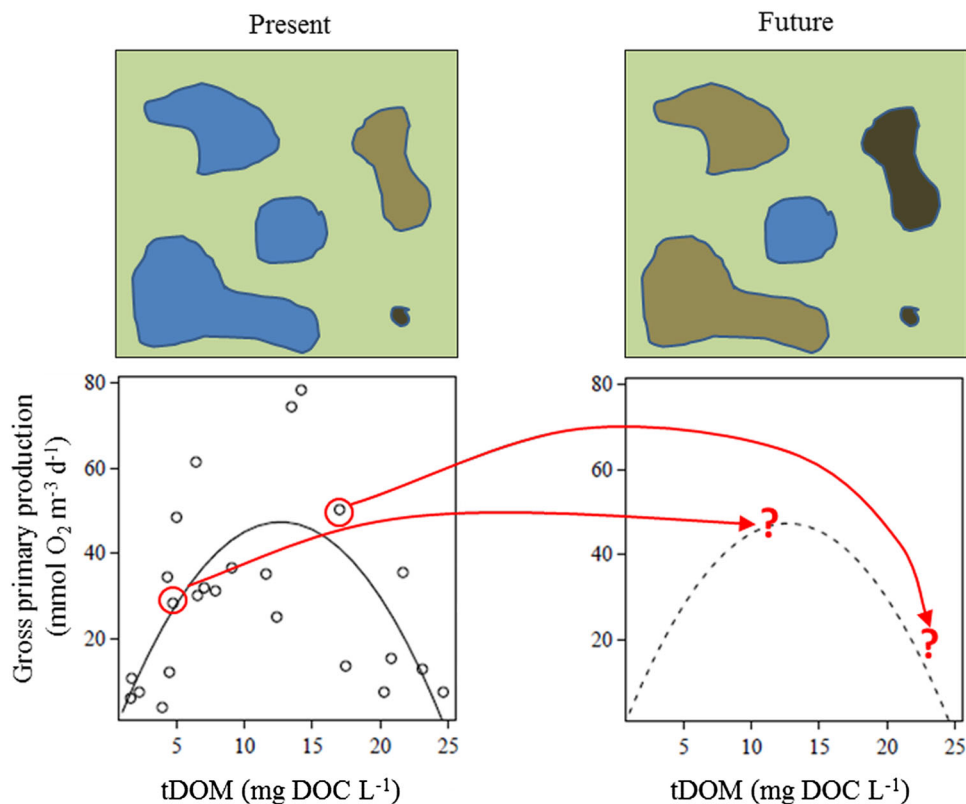


Figure 6. Future changes in tDOM inputs may alter existing relationships on which our understanding of tDOM effects is built. *Left* Present understanding of the relationship between tDOM and phytoplankton gross primary production is built on spatial surveys across existing gradients on the landscape (data from Hanson and others 2003; copyright 2003 by the American Society of Limnology and Oceanography, Inc. and modified by permission of Wiley). *Right* It is unclear whether future changes in tDOM concentration will shift phytoplankton gross primary production along the relationship observed in spatial surveys, or whether those changes will alter the mechanisms that produced the spatial patterns, yielding new relationships.

phytoplankton primary production per unit area shows a hump-shaped relationship with tDOM concentration, with peak rates at intermediate tDOM concentrations (Hanson and others 2003; Figure 6). This pattern seems to occur because hydrologic inputs bring nutrients as well as tDOM, so intermediate tDOM lakes have enough nutrients to support phytoplankton productivity and little enough tDOM that the shading effect on phytoplankton is not strong. Clearer lakes have plenty of light but are nutrient limited, whereas darker lakes have plenty of nutrients but are light limited.

Yet although surveys across space can help develop understanding of mechanisms, they may fail as predictive tools if change across time alters those mechanisms or invokes new ones (Clark and others 2010). For instance, the cross-lake relationship between tDOM and phytoplankton production depends in part on cross-lake correlation between tDOM concentrations and nutrient loads, which may be altered by the changes in soils and hydrology that would drive temporal change in tDOM

within a given lake. The relationship may also depend in part on static lake properties like morphometry that would not change with changing tDOM loads.

Understanding the implications of temporal changes in tDOM inputs will also require that we pay attention to dynamic eco-evolutionary and community assembly processes that have not been apparent given the more static worldview of spatial surveys. Given the many strong effects of tDOM on lake ecosystems, it seems almost certain that changes in tDOM concentrations will impose selective pressures that favor some phenotypes over others. Alternatively, higher tDOM might confer an advantage to fish species that are less reliant on visual feeding, altering community structure over time (Stasko and others 2012). Transient responses like these will likely occur at decadal time scales for fishes, and much more rapidly for invertebrates and microbes with shorter generations and faster dispersal, and may significantly modify ecological responses to changing tDOM loads.

SYNTHESIS AND IMPLICATIONS

Because of the central role that tDOM plays in structuring lake ecosystems and important ecosystem services, resource managers and policymakers need to understand the implications of ongoing and future changes in tDOM inputs to surface waters (Stanley and others 2012). Although aquatic scientists know a great deal about the role of tDOM, new research and new kinds of research are needed to bring our understanding to bear on policy and management.

Quantitative predictions of future tDOM loads and concentrations for particular regions and watersheds will be essential to management decision-making. We know that loads depend on an interacting set of drivers encompassing atmospheric chemistry, hydrology, soil processes, terrestrial vegetation, and climate, and that concentrations depend on loads and the in-lake processing of those loads (Kohler and others 2013). This is a complex system that crosses traditional disciplinary boundaries; furthermore, there are significant uncertainties about future trends in some of the important driving factors. Improving predictions about future loads will require interdisciplinary integration and a variety of approaches including long-term observation, paleolimnological analyses, better sampling in different biomes, conceptual and mechanistic models, and scenario analyses.

To complement those predictions of future loads, we need better understanding of how changes in loads will influence ecosystem processes and services. The non-linear relationships that we have highlighted here indicate that many lakes may be quite resilient to tDOM load alterations, at least with respect to particular processes. On the other hand, the complexity of tDOM-driven physical-chemical-biological interactions, and our uncertainty about the applicability of space-for-time substitutions, suggests that understanding tDOM-driven changes will require research that is both mechanistic and holistic. Observational surveys have been and will continue to be essential, especially when they can describe important thresholds and non-linear relationships. Increasingly, researchers should use replicated experiments and models that can isolate key mechanisms and integrate complex and interacting effects. Whole-ecosystem experiments, although they sacrifice replication, will be essential for understanding the net results of complex interactions. By integrating impact-oriented research with better predictions of future tDOM loads, aquatic scientists will be able to provide policy-relevant science to help maintain

valuable ecosystem services in the face of large-scale change in a major ecosystem driver.

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REFERENCES

- Adin A, Katzhendler J, Alkaslassy D, Rav-Acha D. 1991. Trihalomethane formation in chlorinated drinking water: a kinetic model. *Water Res* 25:797–805.
- Aitkenhead-Peterson J, McDowell W, Neff J, Stuart E, Robert L. 2003. Sources, production, and regulation of allochthonous dissolved organic matter inputs to surface waters. San Diego: Academic Press.
- Ask J, Karlsson J, Persson L, Ask P, Bystrom P, Jansson M. 2009. Terrestrial organic matter and light penetration: effects on bacterial and primary production in lakes. *Limnol Oceanogr* 54:2034–40.
- Barichivich J, Briffa KR, Myneni RB, Osborn TJ, Melvin TM, Ciais P, Piao SL, Tucker C. 2013. Large-scale variations in the vegetation growing season and annual cycle of atmospheric CO₂ at high northern latitudes from 1950 to 2011. *Global Change Biol* 19:3167–83.
- Battin TJ, Luyssaert S, Kaplan LA, Aufdenkampe AK, Richter A, Tranvik LJ. 2009. The boundless carbon cycle. *Nat Geosci* 2:598–600.
- Belyea LR, Malmer N. 2004. Carbon sequestration in peatland: patterns and mechanisms of response to climate change. *Global Change Biol* 10:1043–52.
- Berggren M, Strom L, Laudon H, Karlsson J, Jonsson A, Giesler R, Bergstrom AK, Jansson M. 2010. Lake secondary production fueled by rapid transfer of low molecular weight organic carbon from terrestrial sources to aquatic consumers. *Ecol Lett* 13:870–80.
- Bragée P, Mazier F, Rosén P, Fredh D, Broström A, Granéli W, Hammarlund D. 2013. Forcing mechanisms behind variations in total organic carbon (TOC) concentration of lake waters during the past eight centuries; palaeolimnological evidence from southern Sweden. *Biogeosci Discuss* 10:19969–20003.
- Brett MT, Kainz MJ, Taipale SJ, Seshan H. 2009. Phytoplankton, not allochthonous carbon, sustains herbivorous zooplankton production. *Proc Natl Acad Sci USA* 106:21197–201.
- Brothers S, Köhler J, Attermeyer K, Grossart HP, Mehner T, Meyer N, Scharnweber K, Hilt S. 2014. A feedback loop links

- brownification and anoxia in a temperate, shallow lake. *Limnol Oceanogr* 59:1388–98.
- Canham CD, Pace ML, Papaik MJ, Primack AGB, Roy KM, Maranger RJ, Curran RP, Spada DM. 2004. A spatially explicit watershed-scale analysis of dissolved organic carbon in Adirondack lakes. *Ecol Appl* 14:839–54.
- Carpenter SR, Cole JJ, Kitchell JF, Pace ML. 1998. Impact of dissolved organic carbon, phosphorus, and grazing on phytoplankton biomass and production in experimental lakes. *Limnol Oceanogr* 43:73–80.
- Clark JM, Ashley D, Wagner M, Chapman PJ, Lane SN, Evans CD, Heathwaite AL. 2009. Increased temperature sensitivity of net DOC production from ombrotrophic peat due to water table draw-down. *Global Change Biol* 15:794–807.
- Clark JM, Bottrell SH, Evans CD, Monteith DT, Bartlett R, Rose R, Newton RJ, Chapman PJ. 2010. The importance of the relationship between scale and process in understanding long-term DOC dynamics. *Sci Total Environ* 408:2768–75.
- Clark JM, Chapman PJ, Adamson JK, Lane SN. 2005. Influence of drought-induced acidification on the mobility of dissolved organic carbon in peat soils. *Global Change Biol* 11:791–809.
- Cole JJ, Carpenter SR, Kitchell JF, Pace ML, Solomon CT, Weidel BC. 2011. Strong evidence for terrestrial support of zooplankton in small lakes based on stable isotopes of carbon, nitrogen, and hydrogen. *Proc Natl Acad Sci USA* 108:1975–80.
- Cole JJ, Prairie YT, Caraco NF, McDowell WH, Tranvik LJ, Striegl RG, Duarte CM, Kortelainen P, Downing JA, Middelburg JJ, Melack J. 2007. Plumbing the global carbon cycle: Integrating inland waters into the terrestrial carbon budget. *Ecosystems* 10:171–84.
- Cole L, Bardgett RD, Ineson P, Adamson JK. 2002. Relationships between enchytraeid worms (*Oligochaeta*), climate change, and the release of dissolved organic carbon from blanket peat in northern England. *Soil Biol Biochem* 34:599–607.
- Davidson EA, Janssens IA. 2006. Temperature sensitivity of soil carbon decomposition and feedbacks to climate change. *Nature* 440:165–73.
- De Wit HA, Mulder J, Hindar A, Hole L. 2007. Long-term increase in dissolved organic carbon in streamwaters in Norway is response to reduced acid deposition. *Environ Sci Technol* 41:7706–13.
- del Giorgio PA, Peters RH. 1994. Patterns in planktonic P: R ratios in lakes: influence of lake trophic and dissolved organic carbon. *Limnol Oceanogr* 39:772–87.
- Erlandsson M, Buffam I, Folster J, Laudon H, Temnerud J, Weyhenmeyer GA, Bishop K. 2008. Thirty-five years of synchrony in the organic matter concentrations of Swedish rivers explained by variation in flow and sulphate. *Global Change Biol* 14:1191–8.
- Estlander S, Nurminen L, Olin M, Vinni M, Immonen S, Rask M, Ruuhijarvi J, Horppila J, Lehtonen H. 2010. Diet shifts and food selection of perch *Perca fluviatilis* and roach *Rutilus rutilus* in humic lakes of varying water colour. *J Fish Biol* 77:241–56.
- Evans CD, Chapman PJ, Clark JM, Monteith DT, Cresser MS. 2006. Alternative explanations for rising dissolved organic carbon export from organic soils. *Global Change Biol* 12:2044–53.
- Evans CD, Monteith DT, Cooper DM. 2005. Long-term increases in surface water dissolved organic carbon: observations, possible causes and environmental impacts. *Environ Pollut* 137:55–71.
- Fee EJ, Hecky RE, Kasian SEM, Cruikshank DR. 1996. Effects of lake size, water clarity, and climatic variability on mixing depths in Canadian Shield lakes. *Limnol Oceanogr* 41:912–20.
- Finstad AG, Helland IP, Ugedal O, Hesthagen T, Hessen DO. 2014. Unimodal response of fish yield to dissolved organic carbon. *Ecol Lett* 17:36–43.
- Fork ML, Heffernan JB. 2013. Direct and indirect effects of dissolved organic matter source and concentration on denitrification in northern Florida rivers. *Ecosystems* 17(1):14–28.
- Freeman C, Evans CD, Monteith DT, Reynolds B, Fenner N. 2001a. Export of organic carbon from peat soils. *Nature* 412:785.
- Freeman C, Ostle N, Kang H. 2001b. An enzymic ‘latch’ on a global carbon store—a shortage of oxygen locks up carbon in peatlands by restraining a single enzyme. *Nature* 409:149.
- Frey KE, McClelland JW. 2009. Impacts of permafrost degradation on arctic river biogeochemistry. *Hydrol Process* 23:169–82.
- Futter MN, Butterfield D, Cosby BJ, Dillon PJ, Wade AJ, Whitehead PG. 2007. Modeling the mechanisms that control in-stream dissolved organic carbon dynamics in upland and forested catchments. *Water Res* 43:W02424.
- Geller A. 1986. Comparison of mechanisms enhancing biodegradability of refractory lake water constituents. *Limnol Oceanogr* 31:755–64.
- Gergel SE, Turner MG, Kratz TK. 1999. Dissolved organic carbon as an indicator of the scale of watershed influence on lakes and rivers. *Ecol Appl* 9:1377–90.
- Grey J, Jones RI, Sleep D. 2001. Seasonal changes in the importance of the source of organic matter to the diet of zooplankton in Loch Ness, as indicated by stable isotope analysis. *Limnol Oceanogr* 46:505–13.
- Haaland S, Hongve D, Laudon H, Riise G, Vogt RD. 2010. Quantifying the drivers of the increasing colored organic matter in boreal surface waters. *Environ Sci Technol* 44:2975–80.
- Haei M, Oquist MG, Buffam I, Agren A, Blomkvist P, Bishop K, Lofvenmeyer MO, Laudon H. 2010. Cold winter soils enhance dissolved organic carbon concentrations in soil and stream water. *Geophys Res Lett* 37:L08501.
- Hanson PC, Bade DL, Carpenter SR, Kratz TK. 2003. Lake metabolism: relationships with dissolved organic carbon and phosphorus. *Limnol Oceanogr* 48:1112–19.
- Hanson PC, Carpenter SR, Cardille JA, Coe MT, Winslow LA. 2007. Small lakes dominate a random sample of regional lake characteristics. *Freshw Biol* 52:814–22.
- Hessen DO. 1998. Food webs and carbon cycling in humic lakes. In: Tranvik LJ, Hessen DO, Eds. *Aquatic humic substances: ecology and biochemistry*. Berlin: Springer. p 285–315.
- Hongve D, Riise G, Kristiansen JF. 2004. Increased colour and organic acid concentrations in Norwegian forest lakes and drinking water—a result of increased precipitation? *Aquat Sci* 66:231–8.
- Hope D, Kratz TK, Riera JL. 1996. Relationship between P-CO₂ and dissolved organic carbon in northern Wisconsin lakes. *J Environ Qual* 25:1442–5.
- Houser JN. 2006. Water color affects the stratification, surface temperature, heat content, and mean epilimnetic irradiance of small lakes. *Can J Fish Aquat Sci* 63:2447–55.
- Hughes S, Freeman C, Reynolds B, Hudson JA. 1998. The effects of increased drought frequency on sulphate and dissolved organic carbon in peatland dominated ecosystems. *Proceedings of the 2nd international conference on climate and water*. Helsinki: Edita Limited. pp 311–319.

- Imberger J. 1998. Flux paths in a stratified lake. *Coast Estuar Stud* 54:1–18.
- Jackson TA, Hecky RE. 1980. Depression of primary productivity by humic matter in lake and reservoir waters of the boreal forest zone. *Can J Fish Aquat Sci* 37:2300–17.
- Jansson M, Hickler T, Jonsson A, Karlsson J. 2008. Links between terrestrial primary production and bacterial production and respiration in lakes in a climate gradient in subarctic Sweden. *Ecosystems* 11:367–76.
- Jansson M, Karlsson J, Jonsson A. 2012. Carbon dioxide supersaturation promotes primary production in lakes. *Ecol Lett* 15:527–32.
- Jones RI. 1992. The influence of humic substances on lacustrine planktonic food-chains. *Hydrobiologia* 229:73–91.
- Jones SE, Solomon CT, Weidel BC. 2012. Subsidy or subtraction: how do terrestrial inputs influence consumer production in lakes? *Freshw Rev* 5:37–49.
- Jonsson M, Hedström P, Stenroth K, Hotchkiss ER, Vasconcelos FR, Karlsson J, Bystrom P. 2015. Climate change modifies the size structure of assemblages of emerging aquatic insects. *Freshw Biol* 60(1):78–88.
- Karlsson J, Bystrom P, Ask J, Ask P, Persson L, Jansson M. 2009. Light limitation of nutrient-poor lake ecosystems. *Nature* 460:506–9.
- Karlsson J, Jonsson A, Meili M, Jansson M. 2003. Control of zooplankton dependence on allochthonous organic carbon in humic and clear-water lakes in northern Sweden. *Limnol Oceanogr* 48:269–76.
- Kelly PT, Solomon CT, Weidel BC, Jones SE. 2014. Terrestrial carbon is a resource, but not a subsidy, for lake zooplankton. *Ecology* 95:1236–42.
- Kenny JF, Barber NL, Huston SS, Linsey KS, Lovelace JK, Maupin, MA. 2009. Estimated use of water in the United States in 2005. U.S.G. Survey, editor. p 52.
- Kirk JTO. 1994. Light and photosynthesis in aquatic ecosystems. 2nd edn. Cambridge: Cambridge University Press.
- Kirschbaum MUF. 2006. The temperature dependence of organic-matter decomposition—still a topic of debate. *Soil Biol Biochem* 38:2510–18.
- Kling GW. 1988. Comparative transparency, depth of mixing, and stability of stratification in lakes of Cameroon, West Africa. *Limnol Oceanogr* 33:27–40.
- Kohler SJ, Kothawala D, Futter MN, Liungman O, Tranvik L. 2013. In-lake processes offset increased terrestrial inputs of dissolved organic carbon and color to lakes. *PLoS ONE* 8(8):e70598.
- Krasner SW, Scimmenti MJ, Means EG. 1994. Quality degradation: implications for DBP formation. *J Am Water Works Assoc* 86:34–47.
- Kritzberg ES, Ekstrom SM. 2012. Increasing iron concentrations in surface waters—a factor behind brownification? *Biogeosciences* 9:1465–78.
- Kritzberg ES, Granéli W, Björk J, Brönmark C, Hallgren P, Nicolle A, Persson A, Hansson LA. 2014. Warming and browning of lakes: consequences for pelagic carbon metabolism and sediment delivery. *Freshw Biol* 59:325–36.
- Kullberg A, Bishop KH, Hargeby A, Jansson M, Petersen RC. 1993. The ecological significance of dissolved organic carbon in acidified waters. *Ambio* 22:331–7.
- Lapierre J-F, Guillemette F, Berggren M, del Giorgio PA. 2013. Increases in terrestrially derived carbon stimulate organic carbon processing and CO₂ emissions in boreal aquatic ecosystems. *Nat Commun* 4:2972.
- Larsen S, Andersen T, Hessen DO. 2011a. Climate change predicted to cause severe increase of organic carbon in lakes. *Global Change Biol* 17:1186–92.
- Larsen S, Andersen T, Hessen DO. 2011b. The pCO₂ in boreal lakes: organic carbon as a universal predictor? *Global Biogeochem Cycles* 25(8):1–8.
- Ledesma JLJ, Kohler SJ, Futter MN. 2012. Long-term dynamics of dissolved organic carbon: implications for drinking water supply. *Sci Total Environ* 432:1–11.
- MacIntyre S, Sickman JO, Goldthwait SA, Kling GW. 2006. Physical pathways of nutrient supply in a small, ultraoligotrophic arctic lake during summer stratification. *Limnol Oceanogr* 51:1107–24.
- Matilainen A, Sillanpää M. 2010. Removal of natural organic matter from drinking water by advanced oxidation processes. *Chemosphere* 80:351–65.
- Matilainen A, Vepsäläinen M, Sillanpää M. 2010. Natural organic matter removal by coagulation during drinking water treatment: a review. *Adv Colloid Interface Sci* 159:189–97.
- Matthews B, Mazumder A. 2006. Habitat specialization and the exploitation of allochthonous carbon by zooplankton. *Ecology* 87:2800–12.
- Mattsson T, Kortelainen P, Raika A. 2005. Export of DOM from boreal catchments: impacts of land use cover and climate. *Biogeochemistry* 76:373–94.
- McDonald AT, Mitchell GN, Naden PS, Martin DSJ. 1991. Discoloured water investigations. Report to Yorkshire Water.
- McDonald S, Bishop AG, Prenzler PD, Robards K. 2004. Analytical chemistry of freshwater humic substances. *Anal Chim Acta* 527:105–24.
- McKnight DM, Aiken GR. 1998. Sources and age of aquatic humus. In: Hessen DO, Tranvik LJ, Eds. *Aquatic humic substances*. Berlin: Springer.
- Minor EC, Swenson MM, Mattson BM, Oyler AR. 2014. Structural characterization of dissolved organic matter: a review of current techniques for isolation and analysis. *Environ Sci Process Impacts* 16(9):2064–79.
- Mitchell G, McDonald AT. 1992. Discoloration of water by peat following induced drought and rainfall simulation. *Water Res* 26:321–6.
- Monteith DT, Stoddard JL, Evans CD, de Wit HA, Forsius M, Hogasen T, Wilander A, Skjelkvale BL, Jeffries DS, Vuorenmaa J, Keller B, Kopacek J, Vesely J. 2007. Dissolved organic carbon trends resulting from changes in atmospheric deposition chemistry. *Nature* 450:U537–9.
- Moran MA, Hodson RE. 1990. Bacterial production on humic and nonhumic components of dissolved organic carbon. *Limnol Oceanogr* 35:1744–56.
- Morris DP, Zagarese H, Williamson CE, Balseiro EG, Hargreaves BR, Modenutti B, Moeller R, Queimalinos C. 1995. The attenuation of solar UV radiation in lakes and the role of dissolved organic carbon. *Limnol Oceanogr* 40:1381–91.
- Mulholland PJ. 2003. Large-scale patterns in dissolved organic carbon concentration, flux, and sources. In: Findlay S, Sinsabaugh RL, Eds. *Aquatic ecosystems: interactivity of dissolved organic matter*. Amsterdam: Elsevier.
- Neff JC, Asner GP. 2001. Dissolved organic carbon in terrestrial ecosystems: synthesis and a model. *Ecosystems* 4:29–48.

- Olin M, Vinni M, Lehtonen H, Rask M, Ruuhijärvi J, Saulamo K, Ala-Opas P. 2010. Environmental factors regulate the effects of roach *Rutilus rutilus* and pike *Esox lucius* on perch *Perca fluviatilis* populations in small boreal forest lakes. *J Fish Biol* 76:1277–93.
- Pace ML, Cole JJ, Carpenter SR, Kitchell JF, Hodgson JR, Van de Bogert MC, Bade DL, Kritzberg ES, Bastviken D. 2004. Whole-lake carbon-13 additions reveal terrestrial support of aquatic food webs. *Nature* 427:240–3.
- Palmer ME, Yan ND, Somers KM. 2014. Climate change drives coherent trends in physics and oxygen content in North American lakes. *Clim Change* 124:285–99.
- Pastor J, Solin J, Bridgham SD, Updegraff K, Harth C, Weishampel P, Dewey B. 2003. Global warming and the export of dissolved organic carbon from boreal peatlands. *Oikos* 100:380–6.
- Pérez-Fuentetaja A, Dillon P, Yan N, McQueen D. 1999. Significance of dissolved organic carbon in the prediction of thermocline depth in small Canadian shield lakes. *Aquat Ecol* 33:127–33.
- Prairie YT. 2008. Carbocentric limnology: Looking back, looking forward. *Can J Fish Aquat Sci* 65:543–8.
- Pregitzer KS, Zak DR, Burton AJ, Ashby JA, MacDonald NW. 2004. Chronic nitrate additions dramatically increase the export of carbon and nitrogen from northern hardwood ecosystems. *Biogeochemistry* 68:179–97.
- Read JS, Hamilton DP, Desai AR, Rose KC, MacIntyre S, Lenters JD, Smyth RL, Hanson PC, Cole JJ, Staehr PA, Rusak JA, Pierson DC, Brookes JD, Laas A, Wu CH. 2012. Lake-size dependency of wind shear and convection as controls on gas exchange. *Geophys Res Lett* 39:L09405.
- Read JS, Rose KC. 2013. Physical responses of small temperate lakes to variation in dissolved organic carbon concentrations. *Limnol Oceanogr* 58:921–31.
- Rouillard A, Rosen P, Douglas MSV, Pienitz R, Smol JP. 2011. A model for inferring dissolved organic carbon (DOC) in lake-water from visible-near-infrared spectroscopy (VNIRS) measures in lake sediment. *J Paleolimnol* 46:187–202.
- Roulet N, Moore TR. 2006. Environmental chemistry—browning the waters. *Nature* 444:283–4.
- SanClements MD, Oelsner GP, McKnight DM, Stoddard JL, Nelson SJ. 2012. New insights into the source of decadal increases of dissolved organic matter in acid-sensitive lakes of the Northeastern United States. *Environ Sci Technol* 46:3212–19.
- Schindler DW, Curtis PJ, Bayley SE, Parker BR, Beaty KG, Stainton MP. 1997. Climate-induced changes in the dissolved organic carbon budgets of boreal lakes. *Biogeochemistry* 36:9–28.
- Skjelkvale BL, Stoddard JL, Jeffries DS, Torseth K, Hogasen T, Bowman J, Mannio J, Monteith DT, Mosello R, Rogora M, Rzychon D, Vesely J, Wieting J, Wilander A, Worsztynowicz A. 2005. Regional scale evidence for improvements in surface water chemistry 1990–2001. *Environ Pollut* 137:165–76.
- Sleighter RL, Hatcher PG. 2007. The application of electrospray ionization coupled to ultrahigh resolution mass spectrometry for the molecular characterization of natural organic matter. *J Mass Spectrom* 42:559–74.
- Snucins E, Gunn J. 2000. Interannual variation in the thermal structure of clear and colored lakes. *Limnol Oceanogr* 45:1639–46.
- Sobek S, Algesten G, Bergstrom AK, Jansson M, Tranvik LJ. 2003. The catchment and climate regulation of pCO₂ in boreal lakes. *Global Change Biol* 9:630–41.
- Sobek S, Tranvik LJ, Cole JJ. 2005. Temperature independence of carbon dioxide supersaturation in global lakes. *Global Biogeochem Cycles* 19:GB2003.
- Solomon CT, Bruesewitz DA, Richardson DC, Rose KC, Van de Bogert MC, Hanson PC, Kratz TK, Larget B, Adrian R, Babin BL, Chiu CY, Hamilton DP, Gaiser EE, Hendricks S, Istvánovics V, Laas A, O'Donnell DM, Pace ML, Ryder E, Staehr PA, Torgersen T, Vanni MJ, Weathers KC, Zhu G. 2013. Ecosystem respiration: drivers of daily variability and background respiration in lakes around the globe. *Limnol Oceanogr* 58:849–66.
- Solomon CT, Carpenter SR, Clayton MK, Cole JJ, Coloso JJ, Pace ML, Vander Zanden MJ, Weidel BC. 2011. Terrestrial, benthic, and pelagic resource use in lakes: results from a three-isotope Bayesian mixing model. *Ecology* 92:1115–25.
- Sondergaard M, Phillips G, Hellsten S, Kolada A, Ecke F, Maemets H, Mjelde M, Azzella MM, Oggioni A. 2013. Maximum growing depth of submerged macrophytes in European lakes. *Hydrobiologia* 704:165–77.
- Stanley EH, Powers SM, Lottig NR, Buffam I, Crawford JT. 2012. Contemporary changes in dissolved organic carbon (DOC) in human-dominated rivers: is there a role for DOC management? *Freshw Biol* 57:26–42.
- Stasko AD, Gunn JM, Johnston TA. 2012. Role of ambient light in structuring north-temperate fish communities: potential effects of increasing dissolved organic carbon concentration with a changing climate. *Environ Rev* 20:173–90.
- Steinberg CEW, Kamara S, Prokhotskaya VY, Manusadzianas L, Karasyova TA, Timofeyev MA, Jie Z, Paul A, Meinelt T, Fajalla VF, Matsuo AYO, Burnison BK, Menzel R. 2006. Dissolved humic substances—ecological driving forces from the individual to the ecosystem level? *Freshw Biol* 51:1189–210.
- Striegl RG, Aiken GR, Dornblaser MM, Raymond PA, Wickland KP. 2005. A decrease in discharge-normalized DOC export by the Yukon River during summer through autumn. *Geophys Res Lett* 32:L21413
- Taipale S, Kankaala P, Tiirola M, Jones RI. 2008. Whole-lake dissolved inorganic C-13 additions reveal seasonal shifts in zooplankton diet. *Ecology* 89:463–74.
- Tanentzap AJ, Szokan-Emilson EJ, Kielstra BW, Arts MT, Yan ND, Gunn JM. 2014. Forests fuel fish growth in freshwater deltas. *Nat Commun* 5:9.
- Tanentzap AJ, Yan ND, Keller B, Girard R, Heneberry J, Gunn JM, Hamilton DP, Taylor PA. 2008. Cooling lakes while the world warms: effects of forest regrowth and increased dissolved organic matter on the thermal regime of a temperate, urban lake. *Limnol Oceanogr* 53:404–10.
- Thurman, E. M. 1985. Organic geochemistry of natural waters. Springer.
- Tranvik LJ. 1990. Bacterioplankton growth on fractions of dissolved organic carbon of different molecular weights from humic and clear waters. *Appl Environ Microbiol* 56:1672–7.
- Tranvik LJ. 1998. Degradation of dissolved organic matter in humic waters by bacteria. *Aquatic humic substances*. Berlin: Springer. pp 259–83.
- Tranvik LJ, Bertilsson S. 2001. Contrasting effects of solar UV radiation on dissolved organic sources for bacterial growth. *Ecol Lett* 4:458–63.

- Tranvik LJ, Downing JA, Cotner JB, Loiselle SA, Striegl RG, Ballatore TJ, Dillon P, Finlay K, Fortino K, Knoll LB, Kortelainen PL, Kutser T, Larsen S, Laurion I, Leech DM, McCallister SL, McKnight DM, Melack JM, Overholt E, Porter JA, Prairie Y, Renwick WH, Roland F, Sherman BS, Schindler DW, Sobek S, Tremblay A, Vanni MJ, Verschoor AM, von Wachenfeldt E, Weyhenmeyer GA. 2009. Lakes and reservoirs as regulators of carbon cycling and climate. *Limnol Oceanogr* 54:2298–314.
- Tranvik LJ, Jansson M. 2002. Climate change - Terrestrial export of organic carbon. *Nature* 415:861–2.
- Volk CJ, Volk CB, Kaplan LA. 1997. Chemical composition of biodegradable dissolved organic matter in streamwater. *Limnol Oceanogr* 42:39–44.
- von Lutzow M, Kogel-Knabner I. 2009. Temperature sensitivity of soil organic matter decomposition-what do we know? *Biol Fertil Soils* 46:1–15.
- Weidman PR, Schindler DW, Thompson PL, Vinebrooke RD. 2014. Interactive effects of higher temperature and dissolved organic carbon on planktonic communities in fishless mountain lakes. *Freshw Biol* 59:889–904.
- Weyhenmeyer GA, Prairie YT, Tranvik LJ. 2014. Browning of boreal freshwaters coupled to carbon-iron interactions along the aquatic continuum. *PLoS ONE* 9(2):e88104.
- Williamson CE, Morris DP, Pace ML, Olson AG. 1999. Dissolved organic carbon and nutrients as regulators of lake ecosystems: resurrection of a more integrated paradigm. *Limnol Oceanogr* 44:795–803.
- Williamson CE, Stemberger RS, Morris DP, Frost TM, Paulsen SG. 1996. Ultraviolet radiation in North American lakes: attenuation estimates from DOC measurements and implications for plankton communities. *Limnol Oceanogr* 41:1024–34.
- Winterdahl M, Erlandsson M, Futter MN, Weyhenmeyer GA, Bishop K. 2014. Intra-annual variability of organic carbon concentrations in running waters: drivers along a climatic gradient. *Global Biogeochem Cycles* 28:451–64.
- Wu ZT, Dijkstra P, Koch GW, Penuelas J, Hungate BA. 2011. Responses of terrestrial ecosystems to temperature and precipitation change: a meta-analysis of experimental manipulation. *Global Change Biol* 17:927–42.
- Wüest A, Lorke A. 2003. Small-scale hydrodynamics in lakes. *Ann Rev Fluid Mech* 35:373–412.
- Young KC, Docherty KM, Maurice PA, Bridgman SD. 2005. Degradation of surface-water dissolved organic matter: influences of DOM chemical characteristics and microbial populations. *Hydrobiologia* 539:1–11.