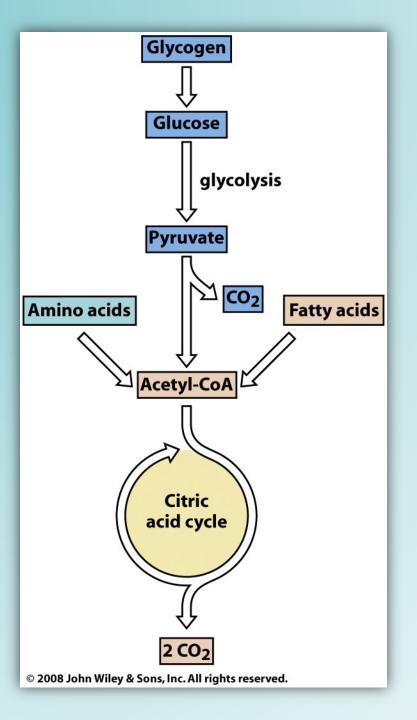
The Tricarboxylic Acid Cycle

(Krebs cycle; citric acid cycle)

CHEM 420 – Principles of Biochemistry Instructor – Anthony S. Serianni

Chapter 21: Voet/Voet, *Biochemistry*, 2011 Fall 2015

November 20 & 23



Integration of glycolysis with TCA

All TCA
reactions are
mitochondrial (matrix);
the PD reaction is
also mitochondrial.

Pvruvate CH₃—C NAD+ pyruvate dehydrogenase CO2 +(NADH COO- Acetyl-CoA CH3-C $c = 0^{H_2O_-}$ †coo-CoASH CH₂ CH₂ — coo-*COO **Oxaloacetate** synthase CH₂ 1/2 COO -NADH +H+ *COO Citrate NAD ‡ COO CH₂ 8. malate 2. aconitase CH₂ 1/2 + COOdehydrogenase -c-coo **L-Malate** HO-C-H7. fumarase *COO_ Isocitrate H₂O Citric acid NAD+ 1/2 TCOO 3. isocitrate cycle dehydrogenase NADH) CH †coo CO2 HC 6. succinate CH₂ 1/2 t COO dehydrogenase **Fumarate** CH, 4. α-ketoglutarate FAD c=0dehydrogenase /2[‡]COO⁻ 5. succinyl-CoA CoASH *COO_ ‡COO- *CO2 synthetase α-Ketoglutarate CH₂ CoASH CH₂ NAD CH₂ CH₂ 1/2 tc00 NADH) c=0Succinate S-CoA Succinyl-CoA © 2008 John Wiley & Sons, Inc. All rights reserved.

Summary of the nine TCA cycle reactions

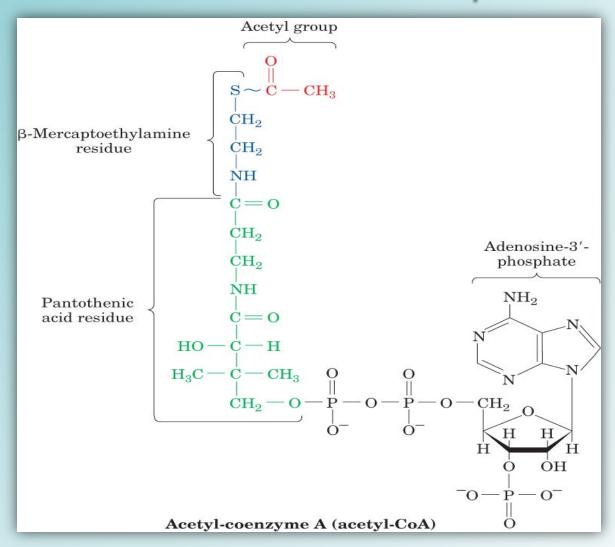
Five reactions produce reducing equivalents (NADH and FADH₂); there is one substrate-level phosphorylation reaction. The C₂ acetyl fragment of acetyl CoA is converted to 2 molecules of CO₂. **Table 17-2**

Standard Free Energy Changes ($\Delta G^{\circ\prime}$) and Physiological Free Energy Changes (ΔG) of Citric Acid Cycle Reactions

Reaction	Enzyme	$\Delta extbf{\emph{G}}^{\circ\prime}$ (kJ \cdot mol $^{-1}$)	∆ <i>G</i> (kJ·mol ^{−1})
1	Citrate synthase	-31.5	Negative
2	Aconitase	~5	~0
3	Isocitrate dehydrogenase	-21	Negative
4	α -Ketoglutarate dehydrogenase	-33	Negative
5	Succinyl-CoA synthetase	-2.1	~0
6	Succinate dehydrogenase	+6	~0
7	Fumarase	-3.4	~0
8	Malate dehydrogenase	+29.7	~0

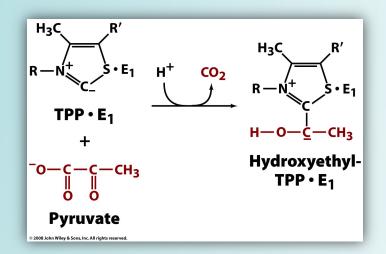
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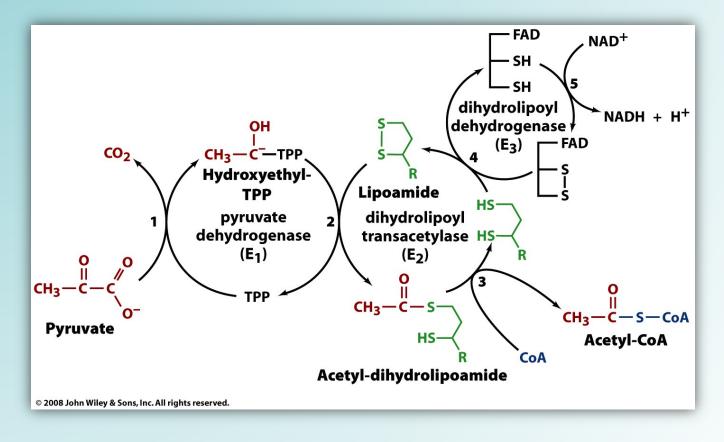
Chemical structure of acetyl-CoA

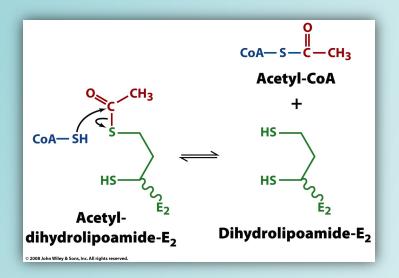


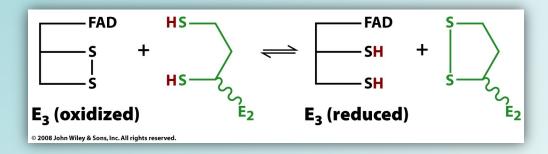
nzymes and Prosthetic Groups of Pyruvate ogenase			
Decarboxylates pyruvate yielding a hydroxyethyl- TPP carbanion			
xyethyl IPP Ip			
group			
nide			
2			
nic			

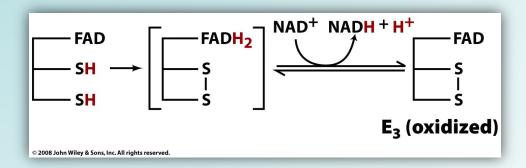
The pyruvate dehydrogenase complex: oxidative decarboxylation of an α -ketoacid (pyruvate)



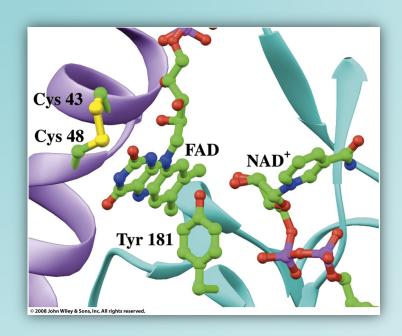


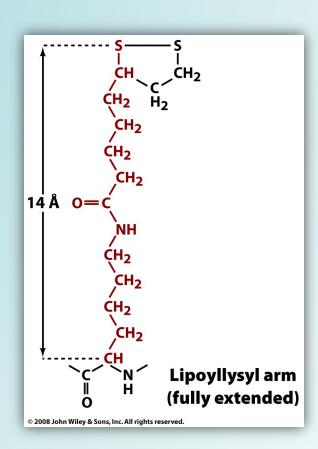






Dihydrolipoyl dehydrogenase mechanism

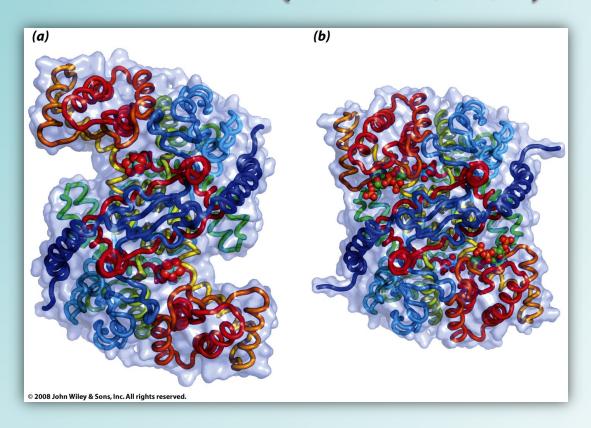




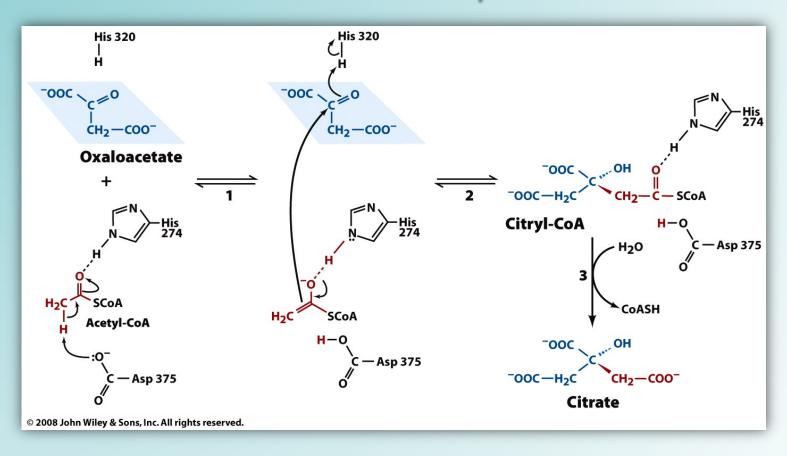
Molecular basis of arsenic poisoning

The pyruvate and α -ketoglutarate dehydrogenase complexes are susceptible to inactivation by arsenic.

Major conformational change occurs when citrate synthase binds its substrate (CS is a homodimer)

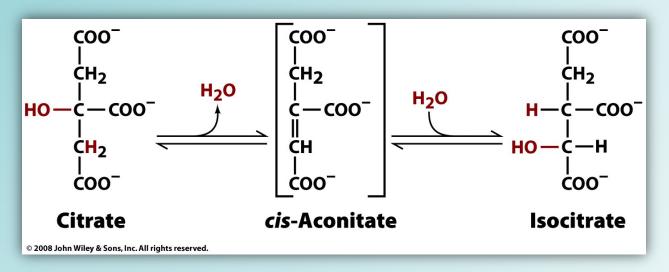


Mechanism of the citrate synthase reaction

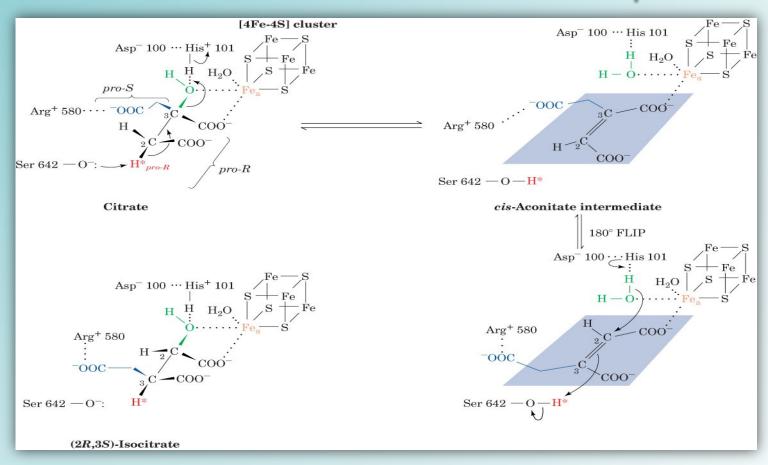


Citrate synthase: reaction stereochemistry

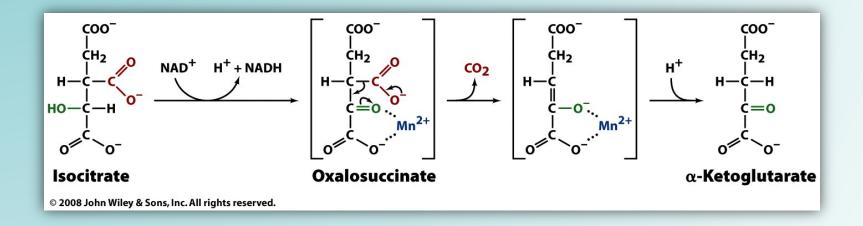
Aconitase



Aconitase reaction stereochemistry

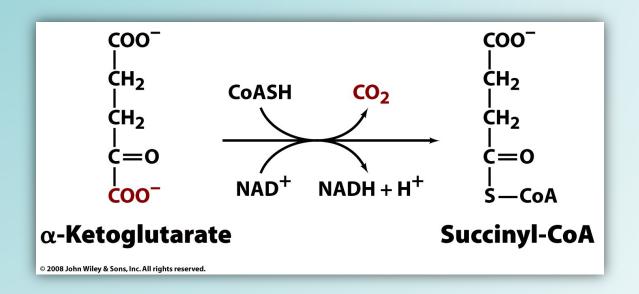


Isocitrate dehydrogenase (NAD+-linked dehydrogenase):



Involves oxidative decarboxylation of a β-ketoacid intermediate

α-Ketoglutarate dehydrogenase (NAD+-linked dehydrogenase):



(mechanism is similar to that of pyruvate dehydrogenase; oxidative decarboxylation of an α -ketoacid)

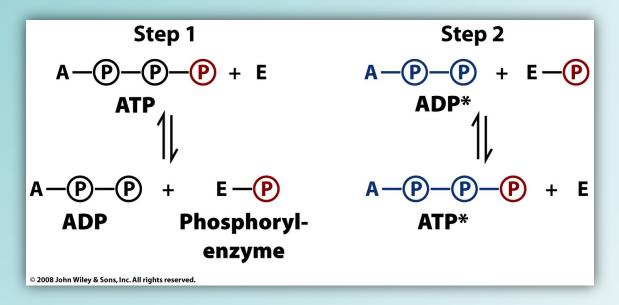
- coo Pyruvate pyruvate dehydrogenase CO2 + NADH COO- Acetyl-CoA CH3-C $c = 0^{H_2O}$ †coo-→ CoASH CH2 CH₂ c — coo-*COO 1. citrate **Oxaloacetate** CH₂ synthase 1/2[†]COO⁻ NADH +H+ *COO Citrate NAD ‡ COO_ CH₂ 8. malate 2. aconitase CH₂ 1/2 tc00 dehydrogenase c — coo-**L-Malate** HO-C-H7. fumarase *COO_ Isocitrate Citric acid NAD 1/2 COO 3. isocitrate cycle dehydrogenase NADH) CH ‡C00 HC CO2 6. succinate CH2 1/2 t COO dehydrogenase **Fumarate** CH, 4. α-ketoglutarate FAD c=0dehydrogenase 1/2 tCOO 5. succinvl-CoA CoASH *COO_ *CO2 ‡c00 synthetase α-Ketoglutarate CH₂ CoASH CH₂ NAD+ CH₂ CH₂ 1/2 t COO NADH) Succinate c=0ZGDP + P, S — CoA Succinyl-CoA © 2008 John Wiley & Sons, Inc. All rights reserved.

Summary of the nine TCA cycle reactions

Five reactions produce reducing equivalents (NADH and FADH₂); there is one substrate-level phosphorylation reaction. The C₂ acetyl fragment of acetyl CoA is converted to 2 molecules of CO₂.

Succinyl CoA synthetase mechanism

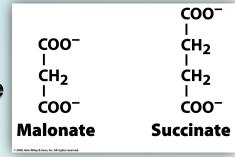
Succinyl CoA synthetase:



Evidence for the involvement of a phosphorylenzyme intermediate; the only substrate-level phosphorylation reaction of the TCA cycle

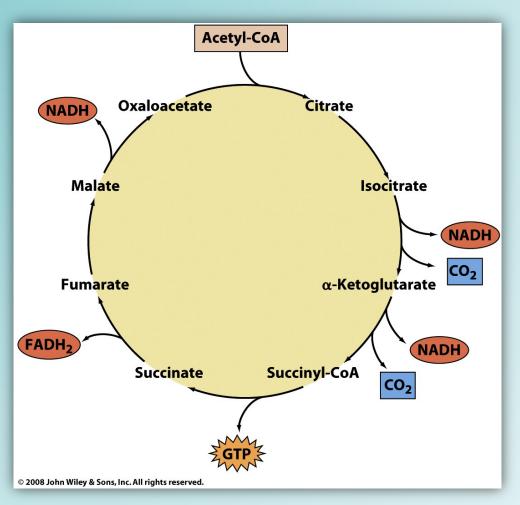
Succinate dehydrogenase (a flavin-linked dehydrogenase):

Inhibited by malonate and succinate



Fumarase:

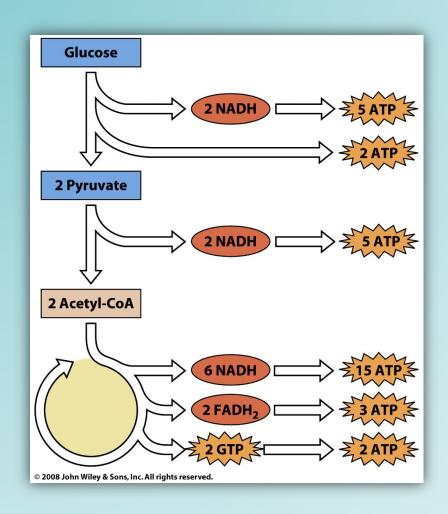
Malate dehydrogenase (NAD+-linked dehydrogenase):



Products generated by the TCA cycle

For each pyruvate entering the cycle:
4 NADH
1 FADH₂
1 GTP

All three carbons of pyruvate are lost as CO₂.



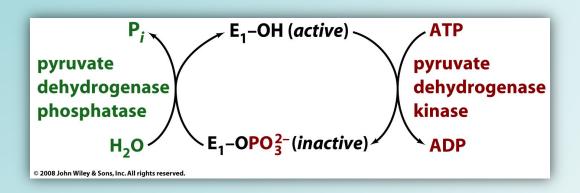
Summary of ATP production from glycolysis, TCA and electron transport/oxidative phosphoryation

2.5 ATP per NADH and 1.5 ATP per FADH₂ entering ET/OP

32 molecules of ATP are produced per glucose under aerobic conditions.

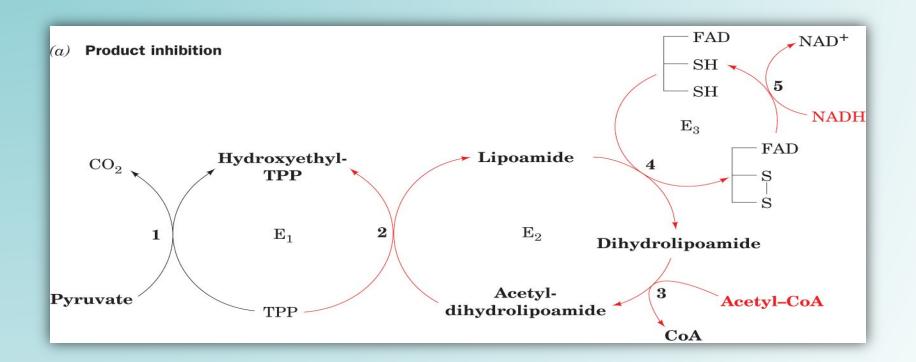
Regulation of the TCA cycle

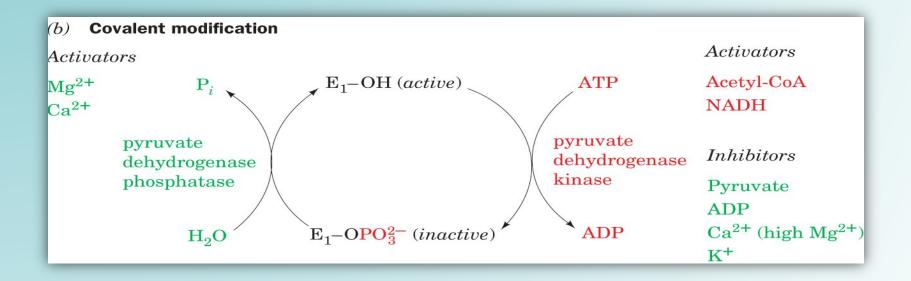
Pyruvate dehydrogenase is regulated by product inhibition (NADH and acetyl CoA) and by covalent modification (phosphorylation of Ser).

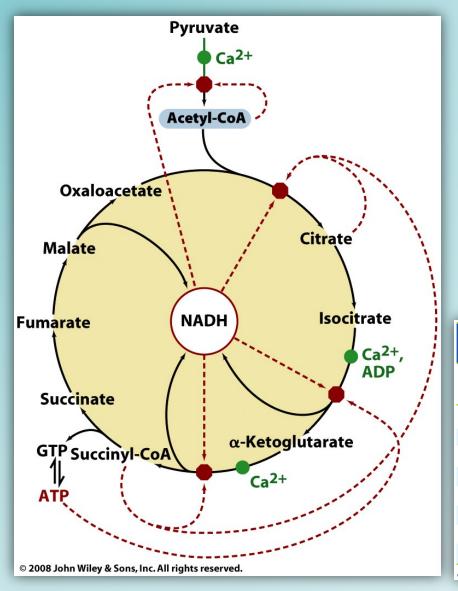


PDK is activated by NADH and acetyl CoA and inhibited by pyruvate, Ca²⁺ and ADP; insulin and Ca²⁺ activate PDP.

PDK and PDP are part of the PD complex.





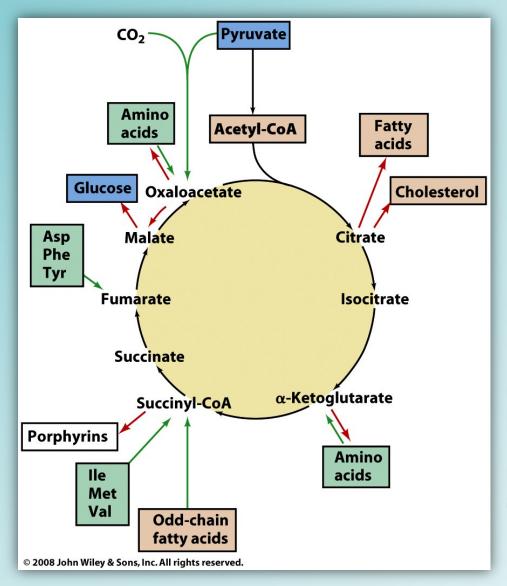


Regulation of the TCA cycle

Red symbols indicate points of inhibition; green symbols are points of activation.

Table 17-2 Standard Free Energy Changes ($\Delta G^{\circ\prime}$) and Physiological Free Energy Changes (ΔG) of Citric Acid Cycle Reactions

synthase se te dehydrogenase	−31.5 ~5 −21	Negative ∼0
e dehydrogenase	21	
ic deliyarogenase	-21	Negative
lutarate dehydrogenase	-33	Negative
I-CoA synthetase	-2.1	~0
te dehydrogenase	+6	~0
se	-3.4	~0
lehydrogenase	+29.7	~0
֡	I-CoA synthetase te dehydrogenase se	I-CoA synthetase -2.1 te dehydrogenase +6 se -3.4 dehydrogenase +29.7



A summary of the amphibolic functions of the TCA cycle, showing cataplerotic (red) and anaplerotic (green) reaction paths

Malate to glucose: gluconeogenesis (transport problem)

Citrate to fatty acids: fatty acid biosynthesis (transport problem)

OAA and αKG to amino acids: amino acid biosynthesis

Conversion of αKG to the amino acid, L-glutamate, by glutamate dehydrogenase (reductive amination)

COO⁻

CH₂

H—C—NH₃⁺

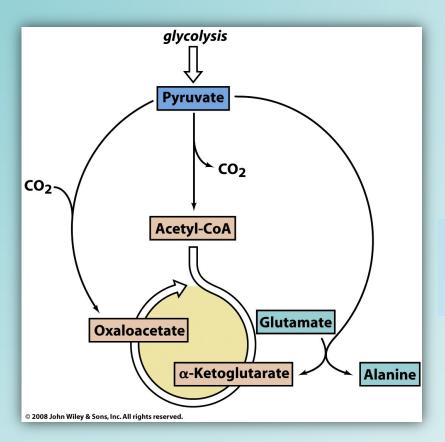
COO⁻

$$\alpha$$
-Ketoglutarate

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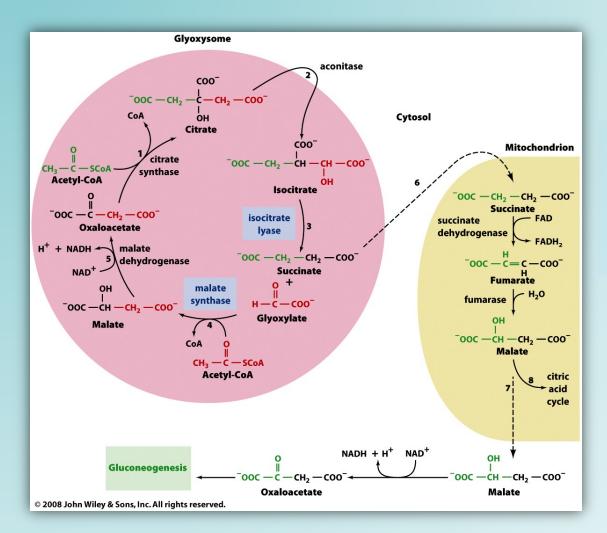
Transamination of OAA with alanine (requires PLP)

COO
$$\stackrel{COO}{\mid}$$
 $\stackrel{COO}{\mid}$ $\stackrel{CH_2}{\mid}$ $\stackrel{CH_2}{\mid}$ $\stackrel{CH_3}{\mid}$ $\stackrel{CH_2}{\mid}$ $\stackrel{COO}{\mid}$ $\stackrel{COO}{\mid$



Ways to replenish TCA intermediates

Pyruvate carboxylase connects glycolysis with TCA directly; PC requires biotin for activity.



Glyoxalate pathway: microorganisms and plants

Allows net conversion
of 2 molecules of acetyl CoA
to OAA instead of 4
molecules of CO₂: valuable
for gluconeogenesis

Organisms that lack the glyoxalate cycle cannot undertake the net synthesis of glucose from acetyl CoA; *i.e.*, they cannot convert fats to carbohydrates.